Synthesis and biological testing of pyrano[2,3-a]acridinones

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The synthesis of 12-amino- 7a and 12-hydroxy-2-arylpyrano[2,3-*a*]acridin-4-ones 7b and 19 *via* the von Strandtmann flavone annelation procedure is described, in which the pyranone ring is formed by the reaction of a β -ketosulfoxide 13, 18 with an aromatic aldehyde.

Introduction

We have recently reported the synthesis and initial biological testing of 12-amino- **7a** and 12-hydroxy-2-phenylpyrano[2,3-*a*]-acridin-4-one **7b**.¹ The synthesis of these pyranoacridinones was achieved by the base-catalysed cyclisation of the imines **5**, followed by aromatisation of the dihydro compounds **6** thus formed (Scheme 1). A key intermediate in this synthetic



sequence was 7-oxo-5,6,7,8-tetrahydroflavone **3** which was condensed with anthranilonitrile **4a**, or ethyl anthranilate **4b**, to give the imines **5**. The oxoflavone **3** was prepared by the acidcatalysed cyclisation of the triketone **2**, itself obtained from the acylation of the diketone **1**.

Although we have used this route to prepare sufficient quantities of 12-amino-2-phenylpyrano[2,3-a]acridin-4-one **7a** for biological testing it has a number of drawbacks; (i) the acylation of the diketone **1** to give the triketone **2** is both capricious and

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poor yielding, (ii) the base-catalysed cyclisation of the imines 5, to the dihydropyrano[2,3-*a*]acridin-4-ones **6**, proceeds in poor yield and (iii) we wished to prepare a number of derivatives with substituted aromatic groups at the 2-position. The synthesis of these derivatives was suggested by our recent review of the structures of known protein tyrosine kinase inhibitors.² Ideally these derivatives would be synthesised from a common, late intermediate in the synthesis.

All of these drawbacks made our original synthetic scheme inappropriate for our purposes and we thus set out to prepare the pyrano[2,3-*a*]acridin-4-ones *via* an alternative route. We report here the elaboration of an acridine skeleton *via* the annelation of a pyranone ring. Crucially, this route involves fewer overall steps, proceeds in higher overall yield and allows the introduction of the aryl group in a simple final step.

Results and discussion

12-Amino-2-phenylpyrano[2,3-*a*]acridin-4-one (APPA) **7a** was prepared in 6 steps, starting from cyclohexane-1,3-dione **8**, by the von Strandtmann flavone annelation procedure,³ a key step in which is the formation of a β -ketosulfoxide by reaction of the dimsyl anion with an ester (Scheme 2). The ester required for



Scheme 2 Reagents and conditions: i, $2-H_2NC_6H_4CN$ 5a, PTSA, PhCH₃, reflux, 68%; ii, K₂CO₃, CuCl, THF, reflux, 60%; iii, NaH (3 equiv.), (EtO)₂CO, 1,2-DME, 15-crown-5, reflux, 32%; iv, DDQ, dioxane, reflux, 81%; v, NaH (4 equiv.), DMSO, THF, 60 °C, 62%; vi, PhCHO, piperidine, PhCH₃, 1,2-DME, DMSO, reflux, 52%

the preparation of 7a is ethyl 9-amino-1-hydroxyacridine-2-carboxylate 12, which we have prepared in four steps from cyclohexane-1,3-dione 8. Cyclohexane-1,3-dione 8 was condensed with anthranilonitrile to give the enamine 9.4 Basecatalysed cyclisation of this enamine, from the most stabilised anion at the 2-position onto the pendant cyano group, followed by tautomerisation, gives the aminoacridinone 10. Ethoxycarbonylation of the anion of this ketone with diethyl carbonate then gave the β -hydroxy ester **11**, which was oxidised to the fully aromatic system 12 using DDQ. The ester 12 was then reacted with the dimsyl anion to give the β -keto sulfoxide **13**. A key feature in the identification of this compound is the presence of an AB system (J14 Hz) for the diastereotopic 2'hydrogens in the ¹H NMR spectrum. Finally, this sulfoxide **13** was reacted with benzaldehyde under base-catalysis to give APPA 7a. The overall yield for this sequence (3.4%) is greater than for our previous synthesis (1.9%), and it also has the advantages that it does not involve the complex and capricious triketone formation, and a range of 2-aryl substituted derivatives can be prepared in the final step, from the sulfoxide 13, by replacing the benzaldehyde with a range of substituted benzaldehydes.

In order to highlight the easy preparation of 2-arylpyrano[2,3-*a*]acridinones using this route, we have prepared a number of 2-aryl-12-hydroxypyrano[2,3-*a*]acridin-4-ones **7b**, **19** (Scheme 3). In this case, the ester **17** required for the prepa-



Scheme 3 Reagents and conditions: i, NaH, DMF, 50 °C then 90 °C, 74%; ii, NaH (3 equiv.), (EtO)₂CO, reflux, 85%; iii, DDQ, dioxane, reflux, 98%; iv, NaH (4 equiv.), DMSO, 70 °C then room temp., 52%; v, ArCHO, piperidine, reflux

ration of 12-hydroxy-2-phenylpyrano[2,3-*a*]acridin-4-one **7b** was prepared in three steps, starting from cyclohexane-1,3-dione **8** and isatoic anhydride **14** (Scheme 3). Initially, the dione and isatoic anhydride were coupled using the method of Mannfred and Siegfried to give 1,2,3,4-tetrahydro-9-hydroxyacridin-1-one **15**.⁵ Acylation of this dione **15** with sodium hydride and diethyl carbonate gave the β -ketoester **16**, which was aromatised with DDQ to give ethyl 1,9-dihydroxyacridine-2-carboxylate **17**. Coupling of the ester with the dimsyl anion, generated from dimethyl sulfoxide and sodium hydride, gave the β -keto sulfoxide **18**, which was reacted with a range of aromatic aldehydes in



Fig. 1 Topoisomerase-DNA cleavable complex

the presence of piperidine to give the 2-aryl-12-hydroxypyrano-[2,3-*a*]acridin-4-ones **7b**, **19**.

12-Amino-2-phenylpyrano[2,3-*a*]acridin-4-one (APPA) **7a** has been tested in the National Cancer Institute (Developmental Therapeutics Program) anti-tumour drug discovery screen. APPA **7a** was shown to inhibit a wide range of cancer cell lines with all IC₅₀s (60 cell lines) in the range 0.1–1.4 µmol dm⁻³. The most promising results were obtained against leukaemia and colon cancer cell lines. As part of this screening process the profile of inhibitory activity is compared to that of anti-tumour agents with known modes of action. The pattern of activity exhibited by APPA **7a** resembles that of known topoisomerase II inhibitors.

The topoisomerases are enzymes which catalyse the topological change of DNA by forming the so-called cleavable complex.⁶ This complex is formed by the topoisomerase II cutting both DNA strands and forming a covalent bond between the tyrosine residue of each protein sub-unit and one of the newly formed 5'-phosphate ends of the DNA, Fig. 1. This process is very similar to that involved in protein tyrosine kinase catalysed phosphorylation, and it is therefore not surprising that APPA **7a**, which we have shown to be a PTK inhibitor, and other PTK inhibitors such as the tyrphostins, *e.g.* AG-555 **20**, also exhibit this activity.⁷



Experimental

Mps were determined on a Gallenkamp apparatus and are uncorrected. Elemental analyses were performed on a Perkin-Elmer 240C. IR Spectra were recorded on a Perkin-Elmer 1600 series FTIR spectrophotometer using sodium chloride plates. ¹H NMR Spectra were acquired on a Bruker WM360 spectrometer at 360 MHz. Coupling constants are given in Hz and all chemical shifts are relative to an internal standard of tetramethylsilane. Low resolution electron impact mass spectra were obtained on a Fisons VG Platform II (Cardiff); high resolution EI and electrospray spectra on a VG ZAB-E spectrometer (EPSRC Mass Spectrometry Service Centre, Swansea). Thin layer chromatography was performed on Merck silica gel 60F254. 1,2-Dimethoxyethane (1,2-DME) and tetrahydrofuran were dried from sodium-benzophenone. 3-(2'-Cyanoanilino)cyclohex-2-en-1-one 9 was prepared by the method of Shutske et al., in 68% yield,⁴ and 1,2,3,4,9,10hexahydroacridine-1,9-dione 15 by the method of Manfred and Seigfried, in 74% yield.⁵

12-Amino-2-phenylpyrano[2,3-a]acridin-4-one 7a

i) 9-Amino-1,2,3,4-tetrahydroacridin-1-one 10. This was prepared *via* a modification of the method of Shutske *et al.*⁴ A solution of 3-(2'-cyanoanilino)cyclohex-2-en-1-one 9 (6.00 g, 0.028 mol), anhydrous K₂CO₃ (8.25 g) and cuprous chloride (0.10 g) was refluxed in dry THF (130 cm³) for a total of 18 h, and filtered while hot into hexane (100 cm³). The resulting precipitate was filtered under suction to give a brown solid, which was recrystallised from propan-2-ol, to give 9-amino-1,2,3,4-tetrahydroacridin-1-one **10** as a pale brown powder (3.56 g, 60%), mp 237–239 °C (Found: C, 73.7; H, 5.5; N, 13.0. C₁₃H₁₂N₂ requires C, 73.6; H, 5.7; N, 13.2%); v_{max} (Nujol)/cm⁻¹ 1605 (C=O); $\delta_{\rm H}$ ([²H₆]DMSO) 2.10 (2 H, q, J6, H-3), 2.70 (2 H, t, J 6, H-2), 3.05 (2 H, t, J 6, H-4), 7.45–7.55 (1 H, m, ArH), 7.60–7.70 (2 H, m, ArH), 8.35–8.40 (1 H, m, ArH) and 8.50 (2 H, br s, NH₂); m/z 212 (M⁺, 41%), 184 (51), 155 (33), 117 (100) and 102 (41).

ii) Ethyl 9-amino-1-hydroxy-3,4-dihydroacridine-2-carboxylate 11. To a sample of freshly washed NaH (0.60 g, 0.015 mol; 60% w/w in mineral oil) was added 1,2-DME (11 cm³) and 15-Crown-5 (5 drops). After stirring for 30 min, 9-amino-1,2,3,4-tetrahydroacridin-1-one 10 (1.11 g, 5.24 mmol) was added and the reaction mixture stirred for a further 30 min. Diethyl carbonate (1.56 cm³) was added slowly to the solution and reflux was maintained for 18 h. On cooling, saturated aqueous NH4Cl was added to neutrality and the resulting precipitate was filtered off to give ethyl 9-amino-1-hydroxy-3,4-dihydroacridine-2-carboxylate 11 as an orange solid (0.48 g, 32%), mp 201-203 °C (Found: M, 284.116. Calc. for 2.70 (2 H, t, J8, H₂-3 or H₂-4), 2.98 (2 H, t, J8, H₂-3 or H₂-4), 4.19 (2 H, q, J7, CH₂CH₃), 7.52 (1 H, t, J8, H-6 or H-7), 7.74 (1 H, t, J8, H-6 or H-7), 7.85 (1 H, d, J8, H-5 or H-8), 8.40 (1 H, d, J8, H-5 or H-8), 9.87 (2 H, br s, NH₂) and 12.50 (1 H, br s, OH); m/z 284 (M⁺, 18%), 243 (100), 238 (41), 221 (46) and 83 (41).

iii) Ethyl 9-amino-1-hydroxyacridine-2-carboxylate 12. Ethyl 9-amino-1-hydroxy-3,4-dihydroacridine-2-carboxylate 11 (0.15 g, 0.53 mmol) and DDQ (0.13 g, 0.58 mmol) were refluxed in 1,4-dioxane (12 cm³) for 12 h. On cooling, the resulting precipitate was filtered off to give ethyl 9-amino-1-hydroxy-acridine-2-carboxylate 12 as a brown powder (0.12 g, 81%), mp >300 °C (Found: M, 282.100. Calc. for C₁₆H₁₃N₂O₃: *M*, 282.100); *v*_{max}(Nujol)/cm⁻¹ 3550 and 3500 (NH₂) and 1730 (C=O); δ_H([²H₆]DMSO) 1.34 (3 H, t, *J*7, CH₃), 4.31 (2 H, q, *J*7, CH₂CH₃), 7.24 (1 H, d, *J*9, H-3 or H-4), 7.46 (1 H, t, *J*8, ArH), 7.64 (1 H, d, *J*8, ArH), 7.90 (1 H, t, *J*8, ArH), 8.23 (1 H, d, *J*9, H-3 or H-4), 8.48 (1 H, d, *J*8, ArH), 10.34 (2 H, br s, NH₂) and 13.00 (1 H, br s, OH); *m*/z 283 (M + 1, 33%), 243 (29), 237 (30), 195 (32) and 149 (25).

iv) 12-Amino-2-phenylpyrano[2,3-a]acridin-4-one 7a. To a sample of freshly washed NaH (0.11 g, 2.75 mmol; 60% w/w in mineral oil) was added anhydrous DMSO (1.88 cm³) and dry THF (5 cm³) and the solution was heated at 60 °C for 45 min under a nitrogen atmosphere. Ethyl 9-amino-1-hydroxyacridine-2-carboxylate 12 (0.20 g, 0.64 mmol) was added dropwise in DMSO (1 cm³) to the reaction mixture, which was left to stir for 1.25 h. Dilute aqueous HCl was added to pH 6, and the solution was extracted with ethyl acetate $(5 \times 100 \text{ cm}^3)$. The combined organic phase was dried and concentrated to 5% volume. 9-Amino-1-hydroxy-2-(2'-methylsulfinylacetyl)acridine 13 precipitated as a green solid (0.14 g, 62%), mp >300 °C; v_{max} - $(Nujol)/cm^{-1} 3490 and 3210 (NH₂), 1637 (C=O) and 1055 (S=O);$ $\delta_{\rm H}([{}^{2}{\rm H}_{6}]{\rm DMSO})$ 2.60 (3 H, s, CH₃), 4.50 (1 H, d, J 14, H-2'), 4.58 (1 H, d, J14, H-2'), 6.40 (1 H, d, J9, H-3 or H-4), 7.40 (1 H, t, J 8, H-6 or H-7), 7.62 (1 H, d, J 8, H-5 or H-8), 7.80 (1 H, t, J 8, H-6 or H-7), 7.99 (1 H, d, J 9, H-3 or H-4), 8.44 (1 H, d, J 8, H-5 or H-8), 9.67 (1 H, br s, OH) and 12.67 (2 H, br s, NH₂); m/z (ES⁻) 314 (20%), 313 (100), 301 (22), 250 (44) and 209 (24).

A mixture of 9-amino-1-hydroxy-2-(2'-methylsulfinylacetyl)acridine **13** (27 mg, 0.09 mmol) and benzaldehyde (20 mg, 0.19 mmol) was refluxed in toluene (7.5 cm³), 1,2-DME (7.5 cm³) and DMSO (0.5 cm³), with a catalytic amount of piperidine (8 drops) over a period of 5 days. On cooling, the solvent was removed and the residue purified by preparative TLC, eluting with ethyl acetate to give 12-amino-2-phenylpyrano[2,3-*a*]acridin-4-one **7a** as an orange solid (15 mg, 52%), mp 310– 312 °C (Found: M, 388.1055. Calc. for C₂₂H₁₄N₂O₂: *M*, 338.1055); v_{max} (Nujol)/cm⁻¹ 3540 and 3400 (NH₂) and 1640 (C=O); $\delta_{\rm H}$ ([²H₆]DMSO) 7.14 (1 H, s, H-3), 7.51 (1 H, t, *J* 8, H-9 or H-10), 7.66–7.72 (3 H, m, H-3', 4', 5'), 7.76 (1 H, d, *J* 9, H-5 or H-6), 7.80 (1 H, t, *J* 8, H-9 or H-10), 7.93 (1 H, d, *J* 8, H-8 or H-11), 8.10 (1 H, d, *J* 9, H-5 or H-6), 8.11–8.17 (2 H, m, H-2', 6'), 8.40 (2 H, br s, NH₂) and 8.55 (1 H, d, *J* 8, H-8 or H-11); *m*/z 339 (M + 1, 44%), 338 (82), 180 (41), 134 (27) and 77 (100).

2-Aryl-12-hydroxypyrano[2,3-a]acridin-4-ones 7b, 19

i) Ethyl 1,2,3,4-tetrahydro-1-oxo-9-hydroxyacridine-2-carboxylate 16. To a sample of freshly washed NaH (0.27 g, 6.75 mmol, 60% w/w in mineral oil) was added dry 1,2-DME (5 cm³) and 15-Crown-5 (5 drops). 1,2,3,4,9,10-Hexahydroacridine-1,9dione 15 (0.50 g, 2.30 mmol) was added and the mixture stirred at room temperature for 30 min then refluxed for 18 h. On cooling, aqueous NH4Cl (10% w/v; 30 cm3) was added and the resulting precipitate filtered off to give ethyl 1,2,3,4-tetrahydro-1-oxo-9-hydroxyacridine-2-carboxylate 16 as a cream powder (0.57 g, 85%), mp 266-267 °C (Found: M, 285.100. Calc. for $C_{16}H_{15}NO_4$: *M*, 285.100); $v_{max}(Nujol)/cm^{-1}$ 3500 (OH) and 1738 (C=O); $\delta_{\rm H}([^{2}{\rm H_{6}}]{\rm DMSO})$ 1.45 (3 H, t, J7, CH₃), 2.45 (2 H, q, J7, H-3), 3.15 (2 H, m, H-4), 3.80 (1 H, t, J7, H-2), 4.35 (2 H, q, J7, CH₂CH₃), 7.55 (1 H, t, J8, H-7), 7.70 (1 H, d, J8, H-5), 7.90 (1 H, t, J8, H-6), 8.25 (1 H, d, J8, H-8) and 12.00 (1 H, br s, OH); m/z 285 (M⁺, 1%), 183 (12), 154 (17), 55 (89), 45 (66) and 43 (100).

ii) Ethyl 1,9-dihydroxyacridine-2-carboxylate 17. A solution of ethyl 1,2,3,4-tetrahydro-1-oxo-9-hydroxyacridine-2-carboxylate 16 (70 mg, 0.21 mmol) and DDQ (63 mg, 0.25 mmol) was refluxed in 1,4-dioxane (5 cm³) for 12 h. On cooling, the resulting precipitate was filtered off to give ethyl 1,9-dihydroxyacridine-2-carboxylate 17 as a brown powder (68 mg, 98%), mp >300 °C (Found: M, 283.0845. Calc. for C₁₆H₁₃NO₄: *M*, 283.0845); *v*_{max}(Nujol)/cm⁻¹ 3500 (OH) and 1705 (CO); $\delta_{\rm H}$ ([²H₆]DMSO) 1.25 (3 H, t, *J* 7, CH₃), 4.20 (2 H, q, *J* 7, CH₂CH₃), 6.95 (1 H, d, *J* 9, H-3 or H-4), 7.40 (1 H, t, *J* 8, ArH), 7.60 (1 H, d, *J* 8, ArH), 7.85 (1 H, t, *J* 8, ArH), 8.05 (1 H, d, *J* 9, H-3 or H-4) and 8.20 (1 H, d, *J* 8, ArH); *m*/*z* 282 (M⁺, 6%), 228 (29), 200 (20), 110 (29), 87 (100) and 77 (73).

iii) 1,9-Dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18. To a sample of freshly washed NaH (0.22 g, 5.5 mmol; 60% w/w in mineral oil dispersion) was added anhydrous DMSO (3.7 cm³) and the solution was heated at 60 °C for 45 min under a nitrogen atmosphere. Ethyl 1,9-dihydroxyacridine-2-carboxylate 17 (0.40 g, 1.41 mmol) was added to the reaction mixture, which was then left to stir for 1.25 h. Aqueous NH₄Cl (15 cm³) was added and the solution extracted with ethyl acetate $(5 \times 100$ cm³). The combined organic phase was dried and concentrated to 5% volume. 1,9-Dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 precipitated as a green solid (0.23 g, 52%), mp 177-179 °C (Found: C, 55.0; H, 5.0; N, 3.55. C₁₆H₁₃NO₄S·(CH₃)₂SO requires C, 55.0; H, 4.8; N, 3.6%); v_{max} (Nujol)/cm⁻¹ 3590 (OH), 1635 (C=O) and 1055 (S=O); $\delta_{\rm H}$ ([²H₆]DMSO) 2.70 (3 H, s, CH₃), 4.55 (1 H, d, J 15, H-2'), 4.66 (1 H, d, J 15, H-2'), 7.05 (1 H, d, J9, H-3 or H-4), 7.45 (1 H, t, J8, H-6 or H-7), 7.70 (1 H, d, J 8, H-5 or H-8), 7.91 (1 H, t, J 8, H-6 or H-7), 8.10 (1 H, d, J 9, H-3 or H-4) and 8.30 (1 H, d, J 8, H-5 or H-8); m/z 316 (M + 1, 100%), 306 (18), 284 (32), 149 (17) and 120 (19).

iv) 12-Hydroxy-2-phenylpyrano[2,3-*a*]acridin-4-one 7b. A mixture of 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 (70 mg, 0.22 mmol) and benzaldehyde (30 mg, 0.24 mmol) was refluxed in toluene (7.5 cm³), 1,2-DME (7.5 cm³) and DMSO

(0.5 cm³), with a catalytic amount of piperidine (6 drops) over a period of 18 h. On cooling, the resluting precipitate was collected to give 12-hydroxy-2-phenylpyrano[2,3-*a*]acridin-4-one **7b** as a yellow solid (43 mg, 57%), mp >300 °C [Found: M, 340.097. Calc. for C₂₂H₁₃NO₃: *M*, 340.097 (M + 1)]; ν_{max} -(Nujol)/cm⁻¹ 1635 (C=O); $\delta_{\rm H}[(^2{\rm H_6}]{\rm DMSO})$ 7.25 (1 H, s, H-3), 7.40 (1 H, t, *J* 8, H-9 or H-10), 7.57 (1 H, d, *J* 9, H-5 or H-6), 7.63–7.64 (4 H, m, H-8 or H-11, H-3', 4', 5'), 7.81 (1 H, t, *J* 8, H-9 or H-10), 8.60 (1 H, d, *J* 9, H-5 or H-6), 8.36 (1 H, d, *J* 8, H-8 or H-11) and 8.55 (2 H, d, *J* 8, H-2', 6'); *m/z* 340 (M + 1, 40%), 260 (60), 220 (71), 167 (86) and 102 (63).

Also prepared by the same method were:

12-Hydroxy-2-(3'-nitrophenyl)pyrano[2,3-a]acridin-4-one

19a. A mixture of 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine (50 mg, 0.16 mmol) and 3-nitrobenzaldehyde (40 mg, 0.26 mmol) in toluene (7.5 cm³), 1,2-DME (7.5 cm³) and DMSO (1 cm³) in the presence of a catalytic amount of piperidine (5 drops) was refluxed over a period of 36 h. On cooling, the resulting precipitate was collected to give 12-hydroxy-2-(3'nitrophenyl)pyrano[2,3-a]acridin-4-one 19a as a green solid (12 mg, 20%), mp >300 °C (Found: M, 384.075. Calc. for $C_{22}H_{12}$ -N₂O₅: *M*, 384.075); v_{max} (Nujol)/cm⁻¹ 1641 (C=O), 1520 and 1349 (NO₂); $\delta_{\rm H}([{}^{2}{\rm H}_{6}]{\rm DMSO})$ 7.35 (1 H, t, J 8, H-9 or H-10), 7.45 (1 H, s, H-3), 7.60-7.70 (3 H, m, H-8 or H-11, H-5 or H-6. OH), 7.80 (1 H, t, J 8, H-9 or H-10), 7.90 (1 H, t, J 9, H-5'), 8.25 (1 H, d, J 9, H-5 or H-6), 8.30 (1 H, d, J 8, H-8 or H-11), 8.45 (1 H, d, J 9, H-6'), 8.90 (1 H, d, J 9, H-4') and 9.45 (1 H, s, H-2'); m/z 384 (M+, 8%), 368 (12), 338 (5), 236 (15) and 77 (100).

12-Hydroxy-2-(4'-nitrophenyl)pyrano[2,3-a]acridin-4-one

19b. From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 (70 mg, 0.22 mmol) and 4-nitrobenzaldehyde (60 mg, 0.44 mmol) in toluene (7.5 cm³), 1,2-DME (7.5 cm³) and DMSO (1 cm³) in the presence of a catalytic amount of piperidine (6 drops) on refluxing for 18 h. On cooling, the resulting precipitate was collected to give 12-hydroxy-2-(4'-nitrophenyl)pyrano-[2,3-a]acridin-4-one **19b** as an orange solid (22 mg, 26%), mp >300 °C [Found: $(M + H)^+$, 385.082. Calc. for $C_{22}H_{12}N_2O_5$: $(M + H)^+$, 385.082]; v_{max} (Nujol)/cm⁻¹ 1637 (C=O), 1338 (NO₂) and 1514 (NO₂); $\delta_{\rm H}$ ([²H₆]DMSO) 7.05 (1 H, s, H-3), 7.30 (1 H, d, J9, H-5), 7.38 (1 H, t, J7.5, H-9 or H-10), 7.58 (1 H, d, J8, H-8 or H-11), 7.75 (1 H, t, J7.5, H-9 or H-10), 7.90 (1 H, d, J9, H-6), 8.26 (1 H, d, J7, H-8 or H-11), 8.34 (2 H, d, J9, H-3' and H-5' or H-2' and H-6'), 8.45 (2 H, d, J9, H-3' and H-5' or H-2' and H-6') and 12.20 (1 H, br s, OH); m/z 385 (M + 1, 36%), 282 (20), 220 (19), 140 (21) and 77 (56).

12-Hydroxy-2-(3'-methoxyphenyl)pyrano[2,3-a]acridin-4-

one 19c. From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 (90 mg, 0.29 mmol) and 3-methoxybenzaldehyde (80 mg, 0.57 mmol) in toluene (20 cm³), DME (20 cm³) and DMSO (1 cm³) with a catalytic amount of piperidine (8 drops) on refluxing for 48 h. On cooling, the resulting precipitate was filtered off to give 12-hydroxy-2-(3'-methoxyphenyl)pyrano-[2,3-a]acridin-4-one 19c as a yellow crystalline solid (55 mg, 52%), mp >300 °C (Found: M, 369.100. Calc. for C₂₃H₁₅NO₄: *M*, 369.100); v_{max} (Nujol)/cm⁻¹ 1636 (C=O); δ_{H} ([²H₆]DMSO) 3.95 (3 H, s, OCH₃), 7.15 (1 H, d, J 8, H-4'), 7.25 (1 H, s, H-3), 7.35 (1 H, t, J8, H-5'), 7.50 (1 H, t, J8, H-9 or H-10), 7.55 (1 H, d, J9, H-5), 7.65 (1 H, d, J8, H-8 or H-11), 7.80 (1 H, t, J 8, H-9 or H-10), 8.05 (1 H, d, J 8, H-6'), 8.15 (1 H, s, H-2'), 8.20 (1 H, d, J9, H-6) and 8.30 (1 H, d, J8, H-8 or H-11); m/z 370 (M + 1, 55%), 275 (42), 207 (30), 180 (15) and 86 (100).

12-Hydroxy-2-(4'-methoxyphenyl)pyrano[2,3-a]acridin-4-

one 19d. From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine **18** (100 mg, 0.32 mmol) and 4-methoxybenzaldehyde (90 mg, 0.63 mmol) in toluene (20 cm³), 1,2-DME (20 cm³) and DMSO (1 cm³) with a catalytic amount of piperidine (8 drops) on refluxing for 48 h. On cooling, the resulting precipitate was filtered off to give 12-hydroxy-2-(4'-methoxyphenyl)pyrano[2,3-*a*]acridin-4-one **19d** as a grey–green powder (89 mg, 76%), mp >300 °C (Found: M, 369.100. Calc. for $C_{23}H_{15}NO_4$: *M*, 369.100); v_{max} (Nujol)/cm⁻¹ 1637 (C=O); δ_H ([²H₆]DMSO) 3.85 (3 H, s, CH₃), 7.10 (1 H, s, H-3), 7.15 (2 H, d, *J* 9, H-3', 5'), 7.35 (1 H, t, *J* 8, H-9 or H-10), 7.60 (1 H, d, *J* 9, H-5), 7.70 (1 H, d, *J* 8, H-8 or H-11), 7.75 (1 H, t, *J* 8, H-9 or H-10), 8.20 (1 H, d, *J* 9, H-6), 8.30 (1 H, d, *J* 8, H-8 or H-11), 8.45 (2 H, d, *J* 9, H-2', 6') and 12.70 (1 H, br s, OH); m/z 369 (M⁺, 6%), 355 (3), 275 (10), 204 (10), 127 (23) and 83 (69).

12-Hydroxy-2-(3'-hydroxyphenyl)pyrano[2,3-*a*]acridin-4-one 19e. From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 (90 mg, 0.29 mmol) and 3-hydroxybenzaldehyde (70 mg, 0.57 mmol) in toluene (20 cm³), 1,2-DME (20 cm³) and DMSO (1 cm³) with a catalytic amount of piperidine (8 drops) on refluxing for 48 h. On cooling, the solvent was removed in vacuo, the residue was dissolved in methanol, and then filtered through Celite. Removal of the solvent gave 12-hydroxy-2-(3'-hydroxyphenyl)pyrano[2,3-a]acridin-4-one 19e as a brown powder (36 mg, 36%), mp >300 °C (Found: M, 355.0845. Calc. for C₂₂H₁₃-NO₄: *M*, 355.0845); v_{max} (Nujol)/cm⁻¹ 1635 (C=O); $\delta_{\text{H}}([{}^{2}\text{H}_{6}]$ -DMSO) 7.00 (1 H, d, J8, H-4'), 7.05 (1 H, s, H-3), 7.30-7.40 (2 H, m, H-9 or H-10 and H-5'), 7.50 (1 H, d, J9, H-5), 7.55 (1 H, d, J8, H-8 or H-11), 7.75 (1 H, t, J8, H-9 or H-10), 7.85 (1 H, s, H-2'), 7.90 (1 H, d, J8, H-6'), 8.20 (1 H, d, J9, H-6), 8.30 (1 H, d, J 8, H-8 or H-11), 9.85 (1 H, s, OH) and 12.25 (1 H, s, OH); m/z 356 (M + 1, 43%), 349 (37), 283 (16), 160 (55), 142 (83) and 86 (100).

12-Hydroxy-2-(4'-hydroxyphenyl)pyrano[**2**,**3**-*a*]**acridin-4-one 19f.** From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine **18** (100 mg, 0.032 mmol) and 4-hydroxybenzaldehyde (0.08 g, 0.63 mmol) in toluene (20 cm³), 1,2-DME (20 cm³) and DMSO (1 cm³) with a catalytic amount of piperidine (8 drops) on refluxing for 48 h. On cooling, the resulting precipitate was collected to give 12-hydroxy-2-(4'-hydroxyphenyl)pyrano[2,3-*a*]acridin-4-one **19f** as a brown solid (33 mg, 30%), mp >300 °C (Found: M, 355.0845. Calc. for C₂₂H₁₃NO₄: *M*, 355.0845); *v*_{max}(Nujol)/cm⁻¹ 1647 (C=O); δ_H([²H₆]DMSO) 7.00 (1 H, s, H-3), 7.05 (2 H, d, J9, H-3', 5'), 7.35 (1 H, t, J8, H-9 or H-10), 7.66 (1 H, d, J8, H-8 or H-11), 7.70–7.80 (2 H, m, H-5 and H-9 or H-10), 8.20 (1 H, d, J9, H-6), 8.33 (1 H, d, J9, H-8 or H-11) and 8.36 (2 H, d, J9, H-2', 6'); *m*/*z* 355 (M⁺, 16%), 107 (62), 90 (97) and 77 (100).

12-Hydroxy-2-(3',4'-dimethoxyphenyl)pyrano[2,3-a]acridin-4-one 19g. From 1,9-dihydroxy-2-(2'-methylsulfinylacetyl)acridine 18 (80 mg, 0.25 mmol) and 3,4-dimethoxybenzaldehyde (80 mg, 0.48 mmol) in toluene (20 cm³), 1,2-DME (20 cm³) and DMSO (1 cm³) in the presence of a catalytic amount of piperidine (8 drops) on refluxing for 36 h. On cooling, the solvent was removed in vacuo, and the residue dissolved in methanol, and filtered through Celite. The solvent was removed in vacuo to give 12-hydroxy-2-(3',4'-dimethoxyphenyl)pyrano-[2,3-a]acridin-4-one 19g as an orange solid (25 mg, 26%), mp >300 °C (Found: M, 399.111. Calc. for C₂₄H₁₇NO₅: M, 399.111); v_{max} (Nujol)/cm⁻¹ 1642 (C=O); $\delta_{\text{H}}([{}^{2}\text{H}_{6}]\text{DMSO})$ 3.85 (3 H, s, OCH₃), 4.0 (3 H, s, OCH₃), 7.15 (1 H, s, H-3), 7.19 (1 H, d, J 8, H-5' or H-6'), 7.35 (1 H, t, J 8, H-9 or H-10), 7.50 (1 H, d, J9, H-5), 7.55 (1 H, d, J8, H-8 or H-11), 7.75 (1 H, t, J 8, H-9 or H-10), 8.10-8.15 (2 H, m, H-2' and H-5' or H-6'), 8.20 (1 H, d, J9, H-6), 8.30 (1 H, d, J8, H-8 or H-11) and 12.25 (1 H, br s, OH); m/z 399 (M⁺, 5%), 237 (24), 151 (86) and 84 (100).

12-Hydroxy-2-(3'-nitro-4'-methoxyphenyl)pyrano[2,3-a]-

acridin-4-one 19h. From 1,9-dihydroxy-2-(2'-methylsulfinyl-acetyl)acridine **18** (100 mg, 0.32 mmol) and 4-methoxy-3-nitrobenzaldehyde (110 mg, 0.64 mmol) in toluene (20 cm³), 1,2-DME (20 cm³) and DMSO (1 cm³) in the presence of a catalytic amount of piperidine (10 drops) on refluxing for 18 h. On cooling, the resulting precipitate was collected to give 12-hydroxy-2-(3'-nitro-4'-methoxyphenyl)pyrano[2,3-*a*]acridin-4-one **19h** as a pale yellow powder (73 mg, 56%), mp >300 °C

(Found: M, 414.085. Calc. for $C_{23}H_{14}N_2O_6$: *M*, 414.085); v_{max} (Nujol)/cm⁻¹ 3259 (OH), 2805 (OMe), 1640 (C=O), 1536 and 1376 (NO₂); δ_{H} ([²H₆]DMSO) 4.03 (3 H, s, CH₃), 7.25 (1 H, s, H-3), 7.35 (1 H, t, *J* 8, H-9 or H-10), 7.58–7.68 (3 H, m, H-5, H-5' and H-8 or H-11), 7.78 (1 H, t, *J* 8, H-9 or H-10), 8.20 (1 H, d, *J* 9, H-6), 8.30 (1 H, d, *J* 8, H-8 or H-11), 8.80 (1 H, d, *J* 8.5, H-6') and 9.00 (1 H, s, 2'-H); *m*/z 415 (M + 1, 35%), 391 (86), 353 (7), 279 (24), 167 (29), 149 (87) and 113 (100).

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